MatSE 584: Point and Line Defects

Fall 2016

Schedule

TR 12:30–2:00pm in 305 MSEB

Course content

• Reading: Google Drive shared folder, articles / e-books through library.illinois.edu
• Additional webnotes: Prof. Helmut Foll’s webnotes and frameless webnotes, hosted locally
• Assignment upload: my.matse.illinois.edu/courses/upload, select MSE 584
• Course website: courses.engr.illinois.edu/mse584; PDF syllabus; Google calendar

Scope

• Point and line defects in crystalline materials: metals, semiconductors, insulators, and ionic materials.
• Quantification and measurement of defect properties: thermodynamic, structural, electronic, characterization.
• Defect interactions.
• Influence of defect properties on macroscale transport (including diffusion), mechanical, and electronic material properties.

Objectives

Students will be able to

(a) critically review the scientific literature;
(b) apply your knowledge to answer scientific questions related to defects;
(c) apply thermodynamics, mechanics, and quantum theory to the properties of defects;
(d) connect defect properties to macroscale material properties;
(e) explain their reasoning to their colleagues in small and large settings;
(f) work together with their colleagues in a professional, scientific manner.

Prerequisites

• Thermodynamics
  – MSE 401/501, CHEM 484, PHYS 427, ME 404 or equivalent
• Electronic Properties of Materials/Condensed Matter Physics
  – MSE 304, PHYS 460/560, or equivalent.
• Introductory Solid Mechanics
  – TAM 251, or equivalent

Instructor

Dallas R. Trinkle (dtrinkle@illinois.edu; 308 MSEB in the west stairwell).

• Associate Professor (joined Univ. Illinois in 2006)
• Computational materials science
  – Crystalline defects (dislocations, point defects, interfaces) from density functional theory
  – Development of new algorithms, computational tools
  – Solid solution softening / strengthening, pipe diffusion, general theory of diffusion

Teaching approach: Team-based learning

An “active learning” approach where we focus on the course objectives: applying knowledge to answer scientific questions about defects and critically engaging with the scientific literature. You will:

• Prepare individually for class (reading papers, background literature, working out conceptual questions);
• Evaluate your team readiness in class with conceptual questions;
• Discuss your questions and deeper points with me and your colleagues;
• Assess the literature, develop consensus with your team, and discuss with other teams.

Team-based learning: Team basics

First off: Teams are not easy.

As you will find out, group work is not always easy. Team members sometimes cannot prepare for or attend group sessions because of other responsibilities, and conflicts often result from differing skill levels and work ethics. When teams work and communicate well, however, the benefits more than compensate for the difficulties. Chances of success are greatly improved if there is an agreement beforehand on what everyone on team expects from everyone else: Team expectations.

In addition, the real world, for the most part, requires people to work together and interface skills, etc. Teamwork is a highly valued skill, but like all skills, requires practice.

1Adapted from Prof. Richard Felder, NCSU.
Team-based learning: Team construction and expectations

• Teams are created for you using CATME by balancing your different skills, viewpoints, etc.
• **Team Expectations.** First meeting of teams will require mutual agreement of team expectations: rules and expectations for everyone in their dealings together as a team. These expectations are for your use and benefit; I will not grade them or even comment on them unless you ask. Be thorough without being unrealistic, as this will give the team the best chance of success:

  * **Realistic:** “We will have read the assigned papers and attempt all prelecture problems before class.” or “We will make sure any who misses meeting for good cause gets caught up on work.”
  * **Unrealistic:** “We will give 110% on every assignment.” Or, “We will read 20 papers in addition to the assigned papers before meeting.” Or, “We will never miss a meeting upon penalty of DEATH.”

Team-based learning: Logistics

The course will cover 8 subtopics in point and line defects. Each week:

• **Monday:** Prelecture questions are due **by midnight.** Done individually, and based on papers and concepts from the papers.
• **Tuesday.** Initial preparation for discussion in class.
  – In-class concept questions that you answer *as a group.*
  – Full class discussion to clear up misconceptions and questions.
  – Team discussions of literature assessment.
• **Wednesday.** Team summary assessment with supporting slides are due **by midnight.**
• **Thursday.** Full class discussion in class.
  – Initial summary of team findings.
  – Full class discussion of papers and team assessments.
  – End of class summary, minute papers

• **End of week:** peer evaluation of team.

Prelecture questions

• Each Monday (starting **Sept. 5**) a few (1-3) prelecture questions will be due at midnight.
• Will be posted on the course website, along with the papers, and suggested background reading.
• Designed to help you check your understanding of concepts in the papers, or used in the papers.
• Will guide opening discussion in Tuesday’s class.
• All assignments are to be submitted electronically as portrait PDF files; they can be uploaded at my.matse.illinois.edu/courses/upload/. Your netid will be automatically appended to the filename, so you should give your file a meaningful, simple name, and can omit adding your own name. If you choose to turn in handwritten homeworks, you will need to scan them and create a PDF (either through a dedicated scanner, or using a free app like JotNot).

Concept quizzes

• Each Tuesday (starting Sept. 6, with practice on Sept. 1) several conceptual questions to check your team understanding.
• Will guide opening discussion in class.
• Multiple choice questions, with partial credit (retry after first attempt).
• Team chooses a weekly “team captain” to lead discussion, arbitrate and manage time. Should rotate weekly.
• Open notes (including prelecture questions).
• Available in-lecture on PrairieLearn.

Literature assessment

The primary driver for our class is engaging actively with scientific literature. This is a primary activity for scientists, and is a skill. It requires

• Reading primary papers, and the papers’ bibliography.
• Working through background reading to build understanding.
• Discussing with others to test your understanding.
• Come to a judgment about:
  – What has been done well/poorly;
  – What experiments might be done to test theories;
  – What modeling might be done to make predictions;
  – What are the outstanding questions?
• In class, your team will summarize and support your assessment in a few Google slides that will drive Thursday discussion.

Minute paper

In the final minutes of Thursday’s class, we will conclude with a minute paper which helps to synthesize your understanding of the lecture, think about your questions, and prime discussion for the next class period. This will be done electronically using the link: illinois.edu/fb/sec/806118. There, you will have three questions:

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2This is what your advisor and future employers (academic, labs, industry) will expect you to be able to do after you graduate.
1. What are the two (or more) most significant (central, useful, meaningful, surprising) things you have learned during this lecture?
2. What main question(s) remain for you?
3. Is there anything that you did not understand?

Following class, I will compile the questions, organize them, and answer some of them online (on the course website). Your responses also help me to adjust the course as needed.

**Grading**

Breakdown determined by majority vote in class:

- **10%** Individual prelecture problems (drop lowest)
- **20%** Team in-class concept questions (drop lowest)
- **60%** Team assessment and discussion of papers (drop lowest)
- **10%** Minute papers (drop lowest)

All participating team members receive the same grade; however, this may be adjusted based on peer evaluation of member contributions over the course of the semester.

**Academic Integrity, Harassment, and Discrimination**

You are bound by the University Honor Code in this course. Any violation of the Honor Code will result in disciplinary action. Please read the university’s academic integrity policies online, especially the section on plagiarism which is not allowed under any circumstances. In addition, harassment or discrimination of any kind will not be tolerated. Please report any concerns immediately to your professor.

**Changes to syllabus**

May occur as deemed necessary by the professor; they will be announced and the updated syllabus posted on the course website.

**Accessing files**

The Univ. Illinois library has access to a huge variety of electronic resources; this plus additional online resources will be our references. Many can be accessed from the library’s website, or via the campus VPN. Alternatively, you can take advantage of the library proxy. This is done by appending proxy2.library.illinois.edu to the web address; when reloaded, you will be asked for Univ. Illinois authentication, and then will be able to access the resource as if you were on campus. In general, this authentication is required only once per session. So, the website
Alternatively, install the Proxy Bookmarklet which makes it extremely easy to use the proxy. I highly recommend this method.

**Google Drive / Google Apps @ Illinois**

In addition, we will use Google Drive to share files. You should have an “infinite” amount of free storage on Google Drive, and you can set up Google Drive so that files are automatically synced to your computer. You may want to upload PDFs from the pre-lecture reading there; if you place these PDFs in the shared folder, please name them `FirstAuthorLastName-Journal-Year.pdf` so that they remain organized. The team slides will be made available using Google Apps.

This means that you will need to either:

- Use your personal Google account to access Google Apps. Be aware that this will count towards your Google storage limitations.
- Activate and use your campus Google account. See the FAQ and more information (including how to create your account).

Your campus Google account will be separate from an existing Google account, should you have one.

**Lecture topics and reading calendar**

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<thead>
<tr>
<th>Tuesday</th>
<th>Thursday</th>
<th>notes</th>
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<tbody>
<tr>
<td>8/23</td>
<td>8/25</td>
<td>intro to MSE584 / concept review</td>
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<tr>
<td>8/30</td>
<td>9/1</td>
<td>practice concept quiz</td>
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<tr>
<td>9/6</td>
<td>9/8</td>
<td><strong>Topic 1</strong>: how many PDs are in a material?</td>
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<td>9/13</td>
<td>9/15</td>
<td><strong>Topic 2</strong>: how do PDs change electronic properties?</td>
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<td>9/20</td>
<td>9/22</td>
<td><strong>no class</strong></td>
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<td>9/27</td>
<td>9/29</td>
<td><strong>Topic 3</strong>: how do PDs react in an open system?</td>
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<td>10/4</td>
<td>10/6</td>
<td><strong>no class</strong></td>
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<tr>
<td>10/11</td>
<td>10/13</td>
<td><strong>no class</strong></td>
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<tr>
<td>Tuesday</td>
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<td>notes</td>
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<td>10/18</td>
<td>10/20</td>
<td><strong>Topic 4:</strong> how do PDs diffuse?</td>
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<td>10/25</td>
<td>10/27</td>
<td><strong>Topic 5:</strong> how do $\perp$s interact in the far-field?</td>
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<td>11/1</td>
<td>11/3</td>
<td><strong>Topic 6:</strong> what is the core structure of a $\perp$?</td>
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<td>11/8</td>
<td>11/10</td>
<td><strong>Topic 7:</strong> how do PDs and $\perp$s interact?</td>
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<td>11/15</td>
<td>11/17</td>
<td><strong>Topic 8:</strong> how do $\perp$s move in a material?</td>
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<tr>
<td>11/22</td>
<td>11/24</td>
<td><strong>Thanksgiving break</strong></td>
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<td>11/29</td>
<td>12/1</td>
<td>no class</td>
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<tr>
<td>12/6</td>
<td>12/8</td>
<td>no class</td>
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**Background reading**

You may want to review the following references to refresh yourself on particular prerequisite topics:

- Ibach, H. and Lueth, H. *Solid-State Physics*. (Springer Berlin Heidelberg: Berlin, Heidelberg, 2010). doi:10.1007/978-3-540-93804-0. There are multiple useful chapters here:
  - Chapters 2 and 3 cover lattices and crystals
  - Chapters 4 and 5 cover phonons
  - Chapters 6 and 7 cover electrons and band structure

- Thermodynamics: chemical potential and mass action law available in Prof. Helmut Foll’s webnotes
  - The chemical potential
  - Mass action law
  - Pitfalls and Extensions of the Mass Action Law
  - Chemical examples for mass action law applications
  - Alternative derivations of the mass action law